

Differing Effects of Supplemental KCl and KHCO_3 : Pathophysiological and Clinical Implications

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Compared to the prehistoric diet, the modern human diet contains not only excessive NaCl and deficient K^+ , but also deficient precursors of HCO_3^- and sometimes excessive precursors of nonvolatile acid. The mismatch between the modern diet and the still ancient biological machinery of humans subtly but chronically disorders their internal milieu, giving rise to a prolonged state of low-grade potassium deficiency and low-grade metabolic acidosis whose severity increases with age. Supplemental KCl cannot redress this mismatch and correct this state. However, the mismatch is redressed and the state corrected by restoring intakes of K^+ and HCO_3^- to levels approaching those in the diet of our prehistoric forebearers, with either fruits and vegetables or with supplemental KHCO_3 . So restored, KHCO_3 can: (1) attenuate hypertension and possibly prevent its occurrence by suppressing the phenomenon of normotensive NaCl -sensitivity, in part by its natriuretic effect; (2) prevent kidney stones by reducing urinary excretion of calcium and increasing urinary excretion of citrate; (3) ameliorate and protect against the occurrence of osteoporosis by increasing the renal retention of calcium and phosphorus, and by suppressing bone resorption and enhancing bone formation; and (4) likely prevent stroke.

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PPOTASSIUM CHLORIDE has long been the preferred potassium salt given to correct hypokalemia in the clinical setting of hypochloremic alkalosis and persisting chloride depletion.¹ However, it is now clear that supplemental K-citrate, can readily correct the modest to moderate hypokalemia induced by hydrochlorothiazide in the relatively small dosages currently used to treat either hypertension or kidney stones,^{2,3} at least in subjects in whom dietary chloride is not severely restricted. K-citrate or KHCO_3 can do so without inducing alkalosis.^{2,3} In fact, over time, KHCO_3 can correct hypochloremic alkalosis and severe hypokalemia when dietary Cl^- is provided at only a modest intake of NaCl .⁴

THE QUESTION

With respect to the optimal choice of a supplemental potassium salt to correct K^+ deficiency, the major public health question today is not whether KCl should be supplemented to expeditiously cor-

rect hypokalemia and an attendant hypochloremic alkalosis. Rather, the major question is whether supplemental KHCO_3 can prevent or treat diseases of major public health importance by correcting the modern diet's generally unrecognized deficiencies of both potassium and bicarbonate (as bicarbonate-yielding organic anions). The deficiencies are generally unrecognized because they are not usually severe enough to cause either hypokalemia or metabolic acidosis, at least as either is traditionally characterized⁵⁻⁸ (see below).

THE MISMATCH

The deficiency of K^+ and HCO_3^- , like the excess of Na^+ in the modern diet, must be considered in the context of human evolution.^{9,10} The diet of our prehistoric human forebearers contained large amounts of K^+ and HCO_3^- -yielding precursors, eg, citrate (from fruits and vegetables, which contain little Cl^-), but little of the then scarce NaCl , 160 to 200 and 40 to 70 mmol per day, respectively. Thus, over millions of years, humans evolved biological machinery to process this prehistoric dietary mix into an internal milieu of electrolytes and alkalinity optimally conducive to biological health. However, in modern times, the mix of minerals in the human diet has changed drastically: K^+ and HCO_3^- -yielding substances are now much lower and Na^+ and Cl^- much higher (Fig). In fact, in humans, the prehistoric dietary K^+/Na^+ ratio has become reversed, and in the United States, the extent of that reversal increases with age.¹¹ Yet, our mineral-processing biological machinery has remained essentially unchanged, ge-

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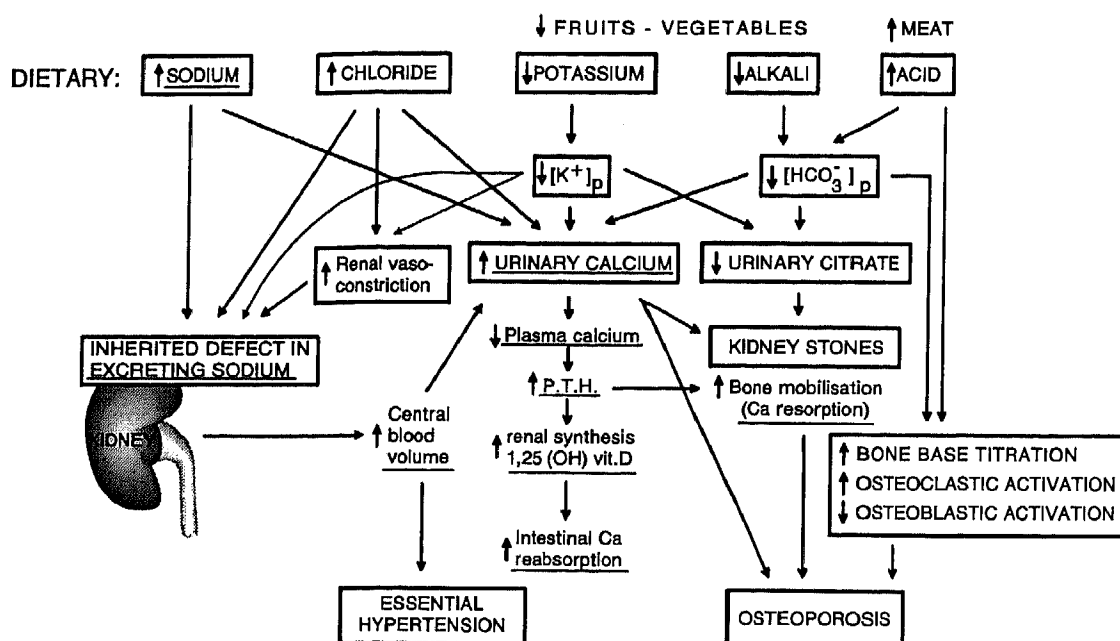


Fig. Hypothesis on the possible links between dietary electrolytes, the kidney, essential hypertension, kidney stones, and osteoporosis: possible roles of excessive dietary sodium, chloride and acid and deficient potassium and alkali. Modified and reprinted with permission from MacGregor and Cappuccio.¹³ The underlined dietary determinants and pathogenetic events are those specified in the hypothesis as originally formulated and depicted by these authors. In the formulation shown here, 'osteoporosis' has replaced the original term 'bone demineralization.'

netically fixed in prehistoric time. Thus, the new mix of minerals in the modern diet is profoundly mismatched to our still ancient biological machinery. By chronically disordering the body's internal milieu, this dietary mismatch contributes critically to the expression of such "maladies of civilization"¹² as hypertension, kidney stones, osteoporosis, and stroke.

THE DISEASES

Hypertension

MacGregor and his colleagues¹³ have long considered Na^+ to be the major electrolytic pathogen of the modern diet. In a recent article entitled "The Kidney and Essential Hypertension: A Link to Osteoporosis?" they propose that in "salt-replete patients with essential hypertension, a genetic defect in the ability of the kidney to excrete sodium" gives rise not only to hypertension but also to hypercalciuria and a negative calcium balance that leads to the occurrence of both kidney stones and "bone demineralization" (Fig).

In accordance with the hypothesis, MacGregor et al¹⁴ showed some time ago that either restriction

of dietary $NaCl$ ¹⁴ or supplemental KCl attenuates essential hypertension.¹⁵ All supplemental potassium salts can have an antihypertensive effect because they natriuretically contract extracellular fluid/blood volume,¹⁶ an expansion of which is the only pathophysiological determinant of either hypertension or hypercalciuria identified in the formulation of MacGregor and Cappuccio.¹³ (Fig). Yet, whereas supplemental $KHCO_3$ and KCl appear to be similarly natriuretic,¹⁷ the two potassium salts have recently been found to induce opposite effects on the hypertension, stroke, nephropathy, and plasma renin activity (PRA) of the stroke-prone spontaneously hypertensive rat (SHRSP), without inducing detectable differences in urinary excretions of Na^+ or K^+ or in body weight or hematocrit.¹⁸

In the SHRSP fed a normal $NaCl$ diet, supplementing dietary K^+ with KCl exacerbated hypertension, whereas supplementing either $KHCO_3$ or potassium citrate (KB/C) attenuated hypertension, when blood pressure (BP) was measured radiotelemetrically, directly, and continually. Supplemental KCl , but not KB/C, induced strokes, which

occurred in all and only those rats in the highest quartiles of both BP and PRA. The level of PRA was abnormally high with KCl but not with KB/C, and varied directly with that of systolic BP as a continuous function across all treatment groups. The severity of angiopathic nephropathy varied directly with the levels of systolic BP and PRA as a continuous function across all treatment groups.¹⁹ These observations show that, with respect to severity of hypertension, frequency of stroke, and severity of nephropathy, the phenotypic expression of the SHRSP is: (1) either increased or not, depending on whether the anionic component of the potassium salt supplemented is or is not Cl^- ; (2) increased by supplementing Cl^- without supplementing Na^+ , and despite supplementing K^+ ; and hence (3) both selectively Cl^- -sensitive and Cl^- -determined. The observations suggest that in the SHRSP selectively supplemented with Cl^- the likelihood of stroke and severity of nephropathy depends on the extent to which both BP and PRA are increased by Cl^- .

These results might have been predicted. Selectively supplementing dietary Cl^- can induce and enhance renal vasoconstriction²⁰ that likely affects the afferent arteriole,²¹ possibly by activating the tubuloglomerular feed-back response,²² which is exaggerated in the stroke-resistant SHR,²³ the strain from which the SHRSP is derived.²⁴ In the SHR and presumably also in the SHRSP, narrowing of the afferent arteriole may give rise to hypertension.²⁵ The extent of that narrowing in the SHR varies directly with the severity of its pharmacologically attenuated hypertension.²⁵ Thus, by augmenting that narrowing in the SHRSP, supplemental KCl might both exacerbate hypertension and increase PRA,¹⁸ angiotensin II, and aldosterone, and thereby induce stroke and exacerbate renal angiopathy,¹⁸ much as might loading dietary Cl^- with NaCl.^{26,27}

In a recent study of patients selected only for essential hypertension, supplementing dietary K^+ with KHCO_3 induced at 8 weeks a significant, persisting attenuation of hypertension compared with placebo, whereas similarly supplemented KCl did not.²⁸ On the basis of a recently completed study of 140 patients selected only for hypertension, we would conclude that supplemental KHCO_3 and KCl at 16 weeks are both effective antihypertensive agents and that KHCO_3 is at least as effective as KCl (unpublished observations).

The question can be asked: Would the hypertension of patients selected for normal or high levels of PRA be more attenuated by supplemental KHCO_3 than KCl? Sealey et al²⁹ have pointed out that even normal levels of PRA are abnormally high in hypertensive patients because hypertension per se should suppress PRA. They have proposed that such unsuppressed levels of PRA reflect an abnormal increased renal release of renin, as might occur with pathological narrowing of only some afferent arterioles. Clearly, such narrowing can occur. Whether its extent and associated hypertension might be more susceptible to attenuation with KHCO_3 than with KCl remains to be determined.

Can supplemental KHCO_3 prevent hypertension? In 41 metabolically controlled studies of 38 healthy normotensive men (24 black men and 14 white men), we recently showed that in most of the blacks but not the whites, salt sensitivity occurred when dietary K^+ was even marginally deficient, 30 mmol per day, but not attended by hypokalemia, as judged by a fasting serum concentration of K^+ of 4.0 mmol/L. However, the pressor effect of NaCl loading was dose-dependently suppressed when dietary K^+ was increased to 70 and 120 mmol/L per day by supplemental KHCO_3 .⁸ Because normotensive salt sensitivity is a likely and possibly common precursor of hypertension,³⁰ such suppression might prevent or delay the occurrence of hypertension, particularly in the many blacks in whom dietary K^+ is deficient.¹¹ Predictably, over the mainly normal range of dietary K^+ studied, the serum concentration of K^+ remained well within the normal range, increasing only minimally when KHCO_3 was supplemented. However, supplemental KHCO_3 dose-dependently reversed and ultimately more than abolished the large hypercalciuric effect of NaCl loading. In those who are salt sensitive, and in whom dietary calcium is suboptimal as in the normotensive subjects studied, dietary replenishment of calcium may reduce BP.³¹ Accordingly, a calcium-retaining effect of KHCO_3 ^{32,33} might have contributed to its reversal of the pressor effect of dietary NaCl. Similarly, the abundant K^+ and HCO_3^- -yielding anions (like citrate) in fruits and vegetables could mediate the hypocalciuric³⁴ and likely calcium-retaining effect of these foods, and thereby contribute both to their antipressor effect and its enhancement by calcium-rich dairy products.³⁵

Kidney Stones

Hypercalciuria is a major risk factor for the development of kidney stones,³⁶ and both phenomena are reported to occur with increased frequency in patients with essential hypertension.³⁷ Dietary NaCl is a major determinant of urinary calcium.³⁸ However, in a recent epidemiological study of 51,529 men conducted prospectively over 4 years, the incidence of symptomatic kidney stones did not correlate with dietary Na⁺, but did correlate strongly and negatively with dietary K⁺ over a broad but normal range, and directly with the dietary intake of meat.³⁹ As noted, dietary K⁺ is derived mainly from foods poor in Cl⁻ but rich in organates, which are converted in vivo to HCO₃⁻. Dietary K⁺ and HCO₃⁻ can both exert important effects on the urinary excretion of calcium (Fig). In normal humans⁴⁰⁻⁴² as well as in patients with essential hypertension,^{41,43} dietary depletion of K⁺ induces increased urinary excretion of calcium. This phenomenon may reflect in part the fact that even minimal dietary depletion of K⁺ evokes a restricted renal capacity to excrete NaCl⁴⁴ that, combined with a continued unrestricted dietary intake of NaCl, can apparently lead to an expanded blood volume and, in consequence, an increased urinary excretion of calcium.^{40,41} However, like hydrochlorothiazide,^{45,46} K⁺ directly and strongly stimulates calcium reabsorption in the distal renal tubule, apparently by enhancing its luminal membrane transport⁴⁶ while at the same time inhibiting Na⁺ transport in this membrane.⁴⁶ Such a dichotomy of effects also appears to occur with KHCO₃. In humans, either supplemental KHCO₃ or administration of chlorothiazide induces an immediate, substantial reduction in urinary excretion of calcium^{32,42,47} while also inducing a natriuresis.^{8,42} HCO₃⁻ appears to exert a hypocalciuric effect in addition to that of K⁺.⁴⁸ In normal humans, KHCO₃ is hypocalciuric and KCl is not.⁴⁹ In a recently completed study of people selected only for essential hypertension, we observed that supplemental KHCO₃, but not KCl, had a significant hypocalciuric effect (unpublished data).

In addition to hypercalciuria, diminished urinary excretion of citrate is a major risk factor for the formation of kidney stones^{36,50} (Fig), in part because urinary citrate normally complexes urinary calcium in a soluble form. Urinary citrate normally inhibits not only the growth of individual crystals

(eg, calcium oxalate) but also crystal agglomeration, the conjoining of multiple crystals into larger masses that become clinical kidney stones.⁵¹ Hypocitraturia occurs with even modest K⁺ deficiency, presumably because it enhances the net renal tubular reabsorption of filtered citrate,^{52,53} possibly by inducing intracellular acidosis in the proximal renal tubule⁵⁴ that may increase both the Na⁺-coupled luminal membrane transport of citrate⁵⁵ and its mitochondrial metabolism.⁵² Under normal physiological circumstances, administration of KHCO₃ (or K⁻ citrate) induces not only a reduction in the urinary excretion of calcium^{42,56} but also an increase in urinary citrate^{47,49,52}. This citraturic effect of K⁻ citrate is not greater than that of KHCO₃.⁴⁹

Decreases in the urinary excretion of citrate, as well as increases in urinary calcium⁵⁷ are predictably induced by high-meat diets⁵¹ because meat and other animal proteins are rich in amino acids that are metabolized to nonvolatile acid. The typical American diet constitutes an acid load of some 1 to 2 mmol/Kg per day. Such acidogenic diets titrate the plasma HCO₃⁻ concentration to a value slightly lower than it would be otherwise, and even small reductions in the plasma HCO₃⁻ concentration can enhance renal reabsorption of citrate and reduce its excretion.⁵²

Osteoporosis

The acid generated from acidogenic diets can titrate bone.⁵⁸ A metabolic bone disorder like that of osteoporosis can be induced in the rat by rendering it chronically, mildly acidotic with NH₄Cl.⁵⁹ Osteoporosis is described in patients with untreated renal tubular acidosis.⁶⁰ In infants and children with classic renal tubular acidosis (type 1), sustained correction of acidosis with alkali as either KHCO₃ or NaHCO₃ corrects osteopenia and an otherwise stunted somatic growth.^{61,62} In 9 patients diagnosed with incomplete (ie, non-acidotic) renal tubular acidosis, supplement K-citrate induced an increase in bone density.⁶³ Reductions of extracellular pH and HCO₃⁻ concentration are potent and independent signals for stimulation of bone resorption and inhibition of bone formation⁶⁴ (Fig). The mammalian skeleton provides a large reservoir of basic calcium salts whose titration mitigates acid-induced reductions in blood pH and bicarbonate concentration, but at the potential cost of diminished bone mass. Even in normal

people, the renal acidification process does not completely dispatch the normal nonvolatile acid load generated endogenously from dietary precursors.^{5,58} As a consequence, (1) the plasma HCO_3^- concentration is titrated to a value slightly lower than would otherwise occur, and (2) some fraction of the endogenous acid load is buffered by basic bone salts such that calcium is chronically mobilized from bone and excreted. Accordingly, over a period of years in the normally aging human, progressive bone loss might result in part from continuing low-grade metabolic acidosis and prolonged titration of alkaline skeletal calcium salts by acid generated from dietary precursors.⁶⁵

Several observations support this formulation. In normal human subjects: (1) blood pH and plasma HCO_3^- concentration decrease slightly but significantly as endogenous acid production is increased over a normal range by dietary manipulation⁵; (2) endogenous acid production often exceeds net acid excretion^{5,58}; and (3) with aging, plasma HCO_3^- decreases, and the serum concentration of Cl^- increases, reflecting the low-grade metabolic acidosis recently shown to attend the age-related decline of normal renal function and acidification capacity.^{4,6} World-wide, the incidence of hip fractures correlates directly with the ingestion of meat protein,⁶⁶ which is, as noted, a major source of endogenous nonvolatile acid. These and other observations and considerations suggest the following hypothesis: In aging humans the "normal" progressive loss of bone mass will be attenuated by KHCO_3 supplemented in an amount sufficient to titrate the nonvolatile acid that is endogenously produced. This hypothesis predicts that this amount of supplemental KHCO_3 in postmenopausal women will (1) enhance the renal reclamation of calcium and phosphorus and thereby promote positive calcium and phosphorus balance, (2) prevent acid titration of calcium-containing base in bone (and plasma HCO_3^-), and thereby (3) diminish bone resorption, enhance bone formation, and increase plasma HCO_3^- concentration.

We³³ recently reported a positive test of this hypothesis in 18 healthy women (aged 51 to 77 years) all of whom were at least 5 years postmenopausal. Under controlled metabolic conditions, the subjects ate a typical whole-food diet with an endogenous acid production rate predictably productive of a positive acid balance (80 to 90 mmol per day per 60 kg body weight). After a control period

of 6 days and a prior prebalance adaptation of 20 days, KHCO_3 (60 to 90 mmol per day per 60 kg body weight) was administered for 16 days as a liquid supplement in divided amounts with meals.

Supplemental KHCO_3 (1) increased the plasma HCO_3^- concentration and pH only slightly, and to values that remained well within the normal range, but virtually abolished net renal acid excretion; (2) improved external balance of calcium and phosphorus because of sustained reductions in their urinary excretion rates; (3) improved phosphorus balance sufficiently in relation to calcium so that retention of calcium as hydroxyapatite would not have been curtailed; (4) reduced the urinary excretion of hydroxyproline, a biochemical marker of the rate of bone resorption; and (5) increased the serum concentration of osteocalcin, an osteoblast-produced protein and biochemical marker of bone formation.

These conclusions were drawn: (1) In postmenopausal women, short-term supplemental KHCO_3 in amounts just sufficient to dissolve endogenous acid production can reduce the rate of bone resorption, increase the rate of bone formation, and reduce bone mass lost in defense of acid-base homeostasis. (2) Dietary supplements of KHCO_3 may be effective in the prevention and treatment of osteoporosis.

Stroke

In a 12-year prospective study of 859 upper-middle-class white subjects, all over 50 years of age, and residing in the contained community of Rancho Bernardo, California, Khaw and Barret-Connor⁶⁷ found that the incidence of stroke death varied inversely with the dietary intake of potassium independent of BP. It could be calculated that increasing dietary potassium by 10 mmol per day within the normal range would reduce the incidence of stroke death by 40%.⁶⁷ In an 8-year prospective study of 43,738 men in the United States, 40 to 75 years of age, Ascherio et al⁶⁸ recently reported a similar inverse relationship between dietary K^+ intake and the occurrence of stroke. The inverse relationship was stronger in hypertensive than in normotensive men. In those taking diuretics, the intake of K^+ as a supplement was also inversely related to risk of stroke. Tobian¹⁰ has emphasized that the incidence of stroke world-wide varies inversely with dietary intake of K^+ . If deficient dietary K^+ and HCO_3^- are major

determinants of the occurrence of both stroke and osteoporosis, one might expect that elderly women with reduced bone density might be at increased risk of stroke. In fact, Browner et al⁶⁹ have recently reported that ambulatory elderly women with low bone density are at increased risk of stroke, and the risk is substantial. Each standard deviation decrease in bone mineral density of the calcaneus and proximal radius was associated with at 1.3-fold increase in stroke. The magnitude of the increase in risk is comparable to that observed for the relation between systolic BP and stroke in the elderly.

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